CALENDAR/COURSE DESCRIPTION
Thermal Systems Design covers the physics and design of energy conversion systems utilized in many engineering systems. The course presents the underlying physics, thermodynamics and energy transfer applied in energy systems design. The topics include:
1) Energy and Work
2) First Law of Thermodynamics and Applications
3) The Second Law of Thermodynamics
4) Entropy and Reversibility
5) Power Systems and Cycles
6) Design Considerations for Energy Systems
7) Energy systems and their environmental applications will be emphasized throughout the course

The course will be taught by means of lectures, tutorials and assignments. Attendance at the tutorials and lectures is mandatory.

PRE-REQUISITES AND ANTI-REQUISITES
Prerequisite(s): None.
Antirequisite(s): ME 2W04

INSTRUCTOR OFFICE HOURS AND CONTACT INFORMATION
Dr. D.R. Novog
NRB 119
novog@mcmaster.ca
ext. 24904
Office Hours: Thursdays – 2:00-3:00pm
Or by appointment

TEACHING ASSISTANT OFFICE HOURS AND CONTACT INFORMATION
Parker Bondi
Tuesday afternoon at 1:00 PM at my office at ETB 303.
Feng Zhou
JHE A401
Thursday, 1 pm - 2 pm
zhouf5@mcmaster.ca
Amr Ibrahim
COURSE OBJECTIVES

By the end of this course, students should be able to:

- Calculate the changes in state as a result of heat or work for ideal closed systems undergoing a reversible process using the first law of thermodynamics.
- Determine the impact of irreversibility on the state change of a substance undergoing a real process using the second law of thermodynamics.
- Calculate the individual process behaviours and cycle performance for simple energy production cycles.
- Use the second law of thermodynamics to assess process efficiency and process performance.
- Understand the important trade-offs between process conditions, cycle efficiency, materials limits and other constraints for energy production systems.

MATERIALS AND FEES

Required Texts:
“Fundamentals of Thermal Fluid Sciences” by Cengal, Cimbala and Turner. The same book will be adopted in EP 3O04 in Year 3. **The course text is MANDATORY since it is required in each tutorial and in the final exam.**

Calculator:
Only the McMaster Standard Calculator will be permitted in tests and examinations. This is available at the Campus Store.

Other Materials:
None.

COURSE OVERVIEW

<table>
<thead>
<tr>
<th>Module</th>
<th>Topic</th>
<th>Readings</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Intro – Properties, States, State Diagrams</td>
<td>2.all</td>
</tr>
<tr>
<td>2</td>
<td>Properties of Pure Substances and Mixtures</td>
<td>4.all</td>
</tr>
<tr>
<td>3</td>
<td>Energy, Heat, and Work</td>
<td>3.all</td>
</tr>
<tr>
<td>4</td>
<td>The 1st Law of Thermodynamics, conservation of mass, and applications</td>
<td>5.all, 6.all</td>
</tr>
<tr>
<td>5</td>
<td>The Second Law of Thermodynamics</td>
<td>7.all</td>
</tr>
<tr>
<td>6</td>
<td>Entropy, Reversibility and Irreversibility</td>
<td>8.1-8.11</td>
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<tr>
<td>7</td>
<td>Ideal and Non-Ideal Power Production Cycles (Rankine and Brayton Cycles, Otto and Diesel Cycles time permitting)</td>
<td>9.1-9.3, 9.7-9.13 (other sections of Chpt9 as time permits)</td>
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<tr>
<td>8</td>
<td>Sustainability Aspects in Power Production (time permitting)</td>
<td>In-class notes</td>
</tr>
</tbody>
</table>
### ASSESSMENT

<table>
<thead>
<tr>
<th>Component</th>
<th>Weight</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tutorial</td>
<td>60</td>
</tr>
<tr>
<td>Final Exam</td>
<td>40</td>
</tr>
<tr>
<td><strong>Total</strong></td>
<td><strong>100%</strong></td>
</tr>
</tbody>
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### ACCREDITATION LEARNING OUTCOMES

Disclaimer: The Learning Outcomes defined in this section are measured for Accreditation purposes only, and will not be taken into consideration in determining a student’s actual grade in the course.

<table>
<thead>
<tr>
<th>Outcomes</th>
<th>Indicators</th>
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</thead>
<tbody>
<tr>
<td>Determine state changes during a thermodynamic process.</td>
<td>1.2</td>
</tr>
<tr>
<td>Apply the first and second law to determine the changes in state due to heat and work and determine cyclic efficiencies</td>
<td>1.3</td>
</tr>
<tr>
<td>Apply assumptions and simplifications to complex thermo processes</td>
<td>3.2</td>
</tr>
<tr>
<td>Solve for ideal and non-ideal behaviour of common power engineering devices</td>
<td>2.1</td>
</tr>
</tbody>
</table>

For more information on Accreditation, please request information from the Department Office (engphys@mcmaster.ca).

### ACADEMIC INTEGRITY

You are expected to exhibit honesty and use ethical behaviour in all aspects of the learning process. Academic credentials you earn are rooted in principles of honesty and academic integrity.

Academic dishonesty is to knowingly act or fail to act in a way that results or could result in unearned academic credit or advantage. This behaviour can result in serious consequences, e.g. the grade of zero on an assignment, loss of credit with a notation on the transcript (notation reads: “Grade of F assigned for academic dishonesty”), and/or suspension or expulsion from the university.

It is your responsibility to understand what constitutes academic dishonesty. For information on the various types of academic dishonesty please refer to the Academic Integrity Policy, located at [http://www.mcmaster.ca/academicintegrity](http://www.mcmaster.ca/academicintegrity)

The following illustrates only three forms of academic dishonesty:

1. Plagiarism, e.g. the submission of work that is not one’s own or for which other credit has been obtained.
2. Improper collaboration in group work.
3. Copying or using unauthorized aids in tests and examinations.
ACADEMIC ACCOMMODATIONS

Students who require academic accommodation must contact Student accessibility Services (SAS) to make arrangements with a Program Coordinator. Academic accommodations must be arranged for each term of study. Student Accessibility Services can be contact by phone at 905.525.9140 ext. 28652 or e-mail at sas@mcmaster.ca. For further information, consult McMaster University's Policy for Academic Accommodation of Students with Disabilities.

NOTIFICATION OF STUDENT ABSENCE AND SUBMISSION OF REQUEST FOR RELIEF FOR MISSED ACADEMIC WORK

1. The McMaster Student Absence Form is a self-reporting tool for Undergraduate Students to report absences DUE TO MINOR MEDICAL SITUATIONS that last up to 3 days and provides the ability to request accommodation for any missed academic work. Please note this tool cannot be used during any final examination period.

2. You may submit a maximum of 1 Academic Work Missed request per term. It is YOUR responsibility to follow up with your Instructor immediately (NORMALLY WITHIN TWO WORKING DAYS) regarding the nature of the accommodation. Relief for missed academic work is not guaranteed.

3. If you are absent for reasons other than medical reasons, for more than 3 days, or exceed 1 request per term you MUST visit the Associate Dean's Office (JHE/H301). You may be required to provide supporting documentation.

4. This form must be submitted during the period of absence or the following day, and is only valid for academic work missed during this period of absence.

5. It is the prerogative of the instructor of the course to determine the appropriate relief for missed term work in his/her course.

6. You should expect to have academic commitments Monday through Saturday but not on Sunday or statutory holidays. If you require an accommodation to meet a religious obligation or to celebrate an important religious holiday, you may submit the Academic Accommodation for Religious, Indigenous and Spiritual Observances (RISO) Form to the Associate Dean's Office. You can find all paperwork needed here: http://www.eng.mcmaster.ca/current/documents.html

NOTICE REGARDING POSSIBLE COURSE MODIFICATION

The instructor and university reserve the right to modify elements of the course during the term. The university may change the dates and deadlines for any or all courses in extreme circumstances. If either type of modification becomes necessary, reasonable notice and communication with the students will be given with explanation and the opportunity to comment on changes. It is the responsibility of the student to check their McMaster email and course websites weekly during the term and to note any changes.

TURNITIN.COM STATEMENT

In this course we will be using a web-based service (Turnitin.com) to reveal plagiarism. Students will be expected to submit their work electronically to Turnitin.com and in hard copy so that it can be checked for academic dishonesty.
Students who do not wish to submit their work to Turnitin.com must still submit a copy to the instructor. No penalty will be assigned to a student who does not submit work to Turnitin.com. All submitted work is subject to normal verification that standards of academic integrity have been upheld (e.g., on-line search, etc.). To see the Turnitin.com Policy, please go to http://www.mcmaster.ca/academicintegrity/.

ON-LINE STATEMENT FOR COURSES REQUIRING ONLINE ACCESS OR WORK

In this course, we will be using Avenue to Learn. Students should be aware that, when they access the electronic components of this course, private information such as first and last names, user names for the McMaster e-mail accounts, and program affiliation may become apparent to all other students in the same course. The available information is dependent on the technology used. Continuation in this course will be deemed consent to this disclosure. If you have any questions or concerns about such disclosure, please discuss this with the course instructor.

REFERENCE TO RESEARCH ETHICS

The two principles underlying integrity in research in a university setting are these: a researcher must be honest in proposing, seeking support for, conducting, and reporting research; a researcher must respect the rights of others in these activities. Any departure from these principles will diminish the integrity of the research enterprise. This policy applies to all those conducting research at or under the aegis of McMaster University. It is incumbent upon all members of the university community to practice and to promote ethical behaviour. To see the Policy on Research Ethics at McMaster University, please go to http://www.mcmaster.ca/policy/faculty/Conduct/ResearchEthicsPolicy.pdf.